Highly Regioselective [2 + 2] Photocycloaddition of Aromatic Aldehydes to Acetylfurans

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Stereospecific and highly regioselective oxetane formation occurs on irradiation of substituted aromatic aldehydes in the presence of 2-acetylfurans.

The [2 + 2] cycloaddition of carbonyl compounds to furan to yield bicyclic oxetanes (*e.g.* 1 in Scheme 1)¹ is an efficient photochemical process. This Paternò-Büchi reaction has been



exploited in synthetic chemistry,^{2,3} with success particularly by Schreiber and his group.⁴ The bicyclic oxetanes which result from photocycloaddition can be transformed by ring opening at the strained acetal carbon.⁵ Alternatively, the dihydrofuran ring can be selectively manipulated, as in the recent photochemical synthesis of oxetanocin, 3,⁶ a nucleoside with antiviral activity against HIV. The recent discovery of other oxetanes with significant anti-viral properties⁷ makes [2 + 2] photocycloaddition a route of potential importance.



The photocycloaddition of aldehydes to furans shows excellent stereoselectivity (>95%) in forming the exo-adduct (e.g. 2),⁸ but does suffer from one disadvantage: in the attack on an unsymmetrically substituted furan, the reaction shows little regioselectivity. Attack of photoexcited benzaldehyde on 2methylfuran, for example, gives the two adducts 4a and 5a in a 1.3:1 ratio^{4e} (Table 1). Although the adducts are difficult to separate, the greater reactivity of the bridgehead-substituted isomer 4a can lead to its selective destruction on silica gel as a means of separation.⁶ Schreiber has attempted to overcome this problem by the use of a bulky trialkylsilyl or trialkylstannyl group at the 2-position of the furan ring, when improved regioselectivities of up to 20:1 can be obtained for attack at the unsubstituted double bond of the furan (Table 1, entries 2, 3),4e yielding dominantly 5. We now report that aromatic aldehydes undergo a remarkably selective photocycloaddition to 2acetylfurans, with bicyclic oxetane formation occurring at the more substituted furan double bond, to give 4.

Our initial investigations showed that the regioselectivity of photocycloaddition of aromatic aldehydes to alkylfurans and furfuryl alcohols or their derivatives was indeed low (Table 1, entries 4–7). For 3-substituted furans or unsymmetrical 2,5-



Fig. 1 Relative amounts of cycloaddition products formed by attack of photoexcited aromatic aldehydes on substituted furans: 6 and 7, benzaldehyde; 8, 2-chloro- or 4-cyano-benzaldehyde; 9, 3-fluoro- (ratio 5.3), 4-methoxycarbonyl-(ratio 3.3) or 4-cyano-benzaldehyde (ratio 2.3)

disubstituted furans, this lack of selectivity was also evident. Fig. 1 shows the relative amounts of cycloaddition products which resulted from attack of benzaldehyde at each double bond of the substituted furans 6 and 7. However, photolysis of a benzene solution of 4-cyanobenzaldehyde and 2-acetylfuran led to a single major adduct, 4b, isolated by column chromatography in 70% yield. Proton NMR spectroscopy of the crude photolysate showed that the regioselectivity of the reaction was at least 20:1 in favour of attack at the acetylbearing double bond; we were unable to detect the alternative regioisomer 5b. A similar selectivity of attack was found using 4-methoxycarbonylbenzaldehyde or terephthaldehyde as the carbonyl components, which gave 4c and 4d in good yield (65– 80%) (Table 1, entries 8–10).

High regioselectivity (>10:1) was also observed for photocycloaddition of 2-chlorobenzaldehyde or 4-cyanobenzaldehyde to 2-acetyl-5-methylfuran **8** (Fig. 1). Interestingly, 3-acetylfurans such as **9** also underwent photocycloaddition to aromatic aldehydes (3-fluoro-, 4-methoxycarbonyl- or 4-cyanobenzaldehyde), but with the reduced selectivity shown in Fig. 1. In these last mentioned examples, adducts corresponding to cycloaddition at the 2,3- and 4,5-double bonds of the furan were easily separated and identified by ¹H and ¹³C NMR spectroscopy. The notable regioselectivity for attack by excited aldehyde on the more substituted double bond of these furans having electron-withdrawing groups was also found for attack

Table 1 Regioisomeric oxetanes produced by photocycloaddition of 2-substituted furans to aromatic aldehydes

En	try 2-Subst	ituted furan	Aldehyde ArCHO	Ratio 4:5	Ref.	
1	$\mathbf{R} = \mathbf{M}$	e	Ar = Ph	1.3:1	a	
2	$\mathbf{R} = \mathbf{Si}$	Me ₃	Ar = Ph	1:2.5	a	
3	$\mathbf{R} = \mathbf{Si}$	Pr ⁱ 3	Ar = Ph	1:>20	а	
4	$\mathbf{R} = \mathbf{C}\mathbf{I}$	H₂ŎH	Ar = Ph	1.5:1	b	
5	$\mathbf{R} = \mathbf{C}\mathbf{I}$	$H_2OSiBu^tMe_2$	Ar = Ph	1:1.1	b	
6	$\mathbf{R} = \mathbf{M}$	e	Ar = 2-Furyl	1:1	b	
7	$\mathbf{R} = \mathbf{M}$	e	$Ar = 4 - CNC_6H_4$	1:2	b	
8	$\mathbf{R} = \mathbf{A}\mathbf{c}$		$Ar = 4 - CNC_6H_4$	>20:1	b	
9	$\mathbf{R} = \mathbf{A}\mathbf{c}$:	$Ar = 4 - MeO_2CC_6H$	$ _{4} > 20:1$	b	
10	$\mathbf{R} = \mathbf{A}\mathbf{c}$;	$Ar = 4 - CHOC_6 H_4$	> 20:1	b	
11	$\mathbf{R} = \mathbf{C}\mathbf{I}$	N	$Ar = 3 - FC_6 H_4$	>10:1	b	

" Ref. 4e. b This work.

on 2-cyanofuran (Table 1, entry 11), but competitive polymerisation of the furan reduced the yield of oxetane 4e to ca. 10% in this case.

Curiously, although the photocycloadditions were successful with electron-withdrawing substituted aromatic aldehydes or even 2-furaldehyde, they were very slow or not observed for electron-donating substituted examples (*e.g.* 4-methoxybenzaldehyde) or benzaldehyde itself. This may possibly be a consequence of triplet energy transfer from the latter (higher energy) triplet donors to the acetylfuran acceptors.⁹ However, a newly found method for metallation of the bicyclic oxetanes¹⁰ could extend the synthetic usefulness of these adducts.

Experimental

Synthesis of exo-1-Acetyl-6-(4'-cyanophenyl)-2,7-dioxabicyclo[3.2.0]hept-3-ene 4b.—A solution of 4-cyanobenzaldehyde (1.0 g, 7.6 mmol) and 2-acetylfuran (2.66 g, 24 mmol) in benzene (85 cm^3) in the presence of sodium carbonate (0.1 g) under nitrogen was irradiated for 4 h, using a 450 W Hanovia medium-pressure mercury arc and Pyrex glassware. Evaporation and chromatography of the residue on silica gel, eluting with 10% acetone-90% light petroleum (b.p. 40-60 $^{\circ}$ C)-0.03% triethylamine, gave the adduct **4b** (70%); $\delta_{\rm H}(270 \text{ MHz};$ CDCl₃) 7.68 (2 H, d, J 8.6, 3'-H and 5'-H), 7.47 (2 H, d, J 8.6, 2'-H and 6'-H), 6.72 (1 H, dd, J 2.7 and 1.2, 3-H), 5.64 (1 H, d, J 3.5, 6-H), 5.53 (1 H, t, J ca. 3, 4-H), 3.92 (1 H, dt, J 1.2, ca. 3 and ca. 3, 5-H) and 2.33 (3 H, s, CH₃CO); δ_c(67.9 MHz; CDCl₃) 198.5 (CO), 149.0 (C-3), 144.4 (C-1'), 132.6 (C-2' and C-6'), 126.1 (C-3' and C-5'), 118.5 (CN), 113.7 (C-1), 112.3 (C-4'), 104.6 (C-4), 89.8 (C-6), 51.4 (C-5) and 23.7 (CH₃).

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